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(54) Title: SULFONAMIDE DERIVATIVES AND THEIR USE

(57) Abstract

Combinatorial libraries are disclosed which are represented by the formula (I): (T'-L)_q(S-C(O)-L'-II' wherein: S is a solid support; T'-L- is an identifier residue; and -L'-II' is a ligand/linker residue. These libraries contain aryl sulfonamides, N-acyl derivatives, and N-substituted pyrrolidines and piperidines of the formula: Y-A-CO-R¹ which are inhibitors of serine proteases and carbonic anhydrase isozymes. They are useful in the treatment of hyper-coagulation disease and ocular diseases such as glaucoma.

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TITLE OF THE INVENTION

SULFONAMIDE DERIVATIVES AND THEIR USE

BACKGROUND OF THE INVENTION

There is interest in methods for the synthesis of large numbers of diverse compounds which can be screened for various 5 possible physiological or other activities. Techniques have been developed in which one adds individual units sequentially as part of the chemical synthesis to produce all or a substantial number of the possible compounds which can result from all the different choices possible at each sequential stage of the synthesis. For these techniques to be 10 successful, it is necessary for the compounds to be amenable to methods by which one can determine the structure of the compounds so made. Brenner and Lerner (PNAS USA 81: 5381-83 (1992)) and WO 93/20242, for example, describe a synthesis wherein oligonucleotides are produced in parallel with and are chemically linked as genetic tags 15 to oligopeptides as the compounds of interest. WO 93/06121 teaches methods for particles-based synthesis of random oligomers wherein identification tags on the particles are used to facilitate identification of the oligomer sequence synthesized. A detachable tagging system is described in Ohlmeyer et al., Proc. Natl. Acad. Sci. USA, 90, 10922-20 10926, Dec. 1993.

SUMMARY OF THE INVENTION

The present invention relates to combinatorial chemical libraries of compounds encoded with tags and to the use of these libraries in assays to discover biologically active compounds. The present invention also relates to libraries containing aryl sulfonamides, N-acyl derivatives, and N-substituted pyrrolidines and piperidines and using these libraries to identify biologically active members by screening for inhibition of serine proteases such as thrombin and Factor Xa; metallo proteases such as angiotensin converting enzyme (ACE); and carbonic anhydrase isozymes. The present invention also relates to members of the library which are inhibitors of carbonic anhydrase and

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thrombin. The invention also relates to methods for their preparation, intermediates, and to methods and pharmaceutical formulations for using these aryl sulfonamides, N-acyl derivatives, and N-substituted pyrrolidines and piperidines in the treatment of mammals, especially humans.

Because of their activity as inhibitors of serine proteases and carbonic anhydrase isozymes, the compounds of the present invention are useful in the treatment of hypercoagulation disease, post-myocardial infarction treatment, post-angioplasty treatment, and ocular diseases such as glaucoma.

DETAILED DESCRIPTION OF THE INVENTION

The combinatorial libraries of the present invention are represented by Formula I:

$$(T'-L)_{q^-}$$
 S $-C(O)-L'-II'$

15 wherein:

S is a solid support;

T'-L- is an identifier residue;

-L'-II' is a ligand/linker residue; and

q is 3-30.

Preferred compounds of Formula I are those wherein:

T'-L- is of the Formula:

$$-CH_2 O-(CH_2)_n O-Ar O-Ar$$

wherein n = 3-12 when Ar is pentachlorophenyl and n = 4-6 when Ar is 2,4,6-trichlorophenyl;

25 q is 3-13; and -L'- is

wherein the left-hand bond as shown is the point of attachment to the solid support and the right hand bond is the point of attachment to the ligand, and B is O or NH, with the proviso that in (b) B is NH when attached to a carbonyl group in the ligand.

Depending on the choice of L' (see Table 1), the ligands of Formula II may be detached by photolytic, oxidative, or other cleavage techniques. For example, when -L'- is (a) and B is O, photolytic detachment may be represented by:

I
$$\xrightarrow{hv}$$
 (T'-L)_q \xrightarrow{s} C(O)-L" + CO₂ + II'BH

wherein L" is the residue from L' and II'BH is II.

Therefore, compounds of the present invention are also represented by Formula II

wherein:

R¹ is $-N(R^6)-(CH_2)_{2-5}-Z-(CH_2)_{2-5}-R^9$, $-NH(CH_2)_{2-5}-R^9$,

R² is the residue on the α carbon of methionine, O-t-butyl-serine, serine, S-trityl-cysteine, cysteine, aspartic acid-β-t-butyl ester, aspartic acid, glutamic acid-γ-t-butyl ester,

glutamic acid, Nim-trityl-histidine, histidine, NE-Boclysine, lysine, Ng-Mtr-arginine, arginine, N-β-tritylasparagine, asparagine, N-γ-trityl-glutamine, glutamine, Nin-Boc-tryptophan, tryptophan, isoleucine, phenylalanine, glycine, alanine, valine, or leucine;

R³ is

lower alkyl or -(CH₂)_m-Q-X;

R4 is

-Q(R⁷, R⁸)-SO₂NH₂, -(CH₂)_m-R¹⁰, with the proviso that when m = 0, R^{10} is not OH, lower alkyl, a 6-membered aromatic heterocyclic ring containing 1 or 2 N atoms, heteroaryl-lower alkyl,

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$$-(CH_2)_m$$
 , $-(CH_2)_m$ X_{0-1} , X_{0-1} , X_{0-1}

$$X_{0-1}$$
 , or CH_2 N O H O H O H

R⁵ is

lower alkyl, lower cycloalkyl, alkenyl, alkynyl, a monoor bicyclic 6- to 10-membered aromatic ring system, or a mono- or bicyclic 5- to 10-membered heteroaromatic ring system containing 1 or 2 N atoms, either system unsubstituted or substituted with 1-2 substituents selected from halogen, alkoxy, alkyl, CF3, CN, -N(lower alkyl)2, and acylamino;

20 R6 is

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H or lower alkyl;

R⁷, R⁸ is

each independently H, halogen, lower alkyl, alkoxy, CN, -NO₂, -CO-lower alkyl, -N(lower alkyl)₂, or

NH-CO-lower alkyl;

R⁹ is

OH, CONH₂, or COOH;

25 R10 is alkoxy, OH, or COOH;

m is

0-6;

A is -NH-CHR²-, -NH(CH₂)₂₋₁₂-, -N , -N , or the descarboxy residue of a primary or secondary amino acid;

Q is a 5- or 6-membered aromatic or heteroaromatic ring containing 0-3 heteroatoms selected from O, N, and S, or a bicyclic 9- or 10-membered aromatic or heteroaromatic ring system containing 0-3 heteroatoms selected from O, N, and S;

X is

H, lower alkyl, halogen, alkoxy, CF₃, CN, -NO₂,

-CO-lower alkyl, -N(lower alkyl)₂, NH-CO-lower alkyl,
or COOH;

Y is -SO₂R³, -COR⁴, -CO-CH(R²)-NHSO₂R³, -CO-CH(R²)-NHCOR⁴, -CO-NHR⁵, or -COOR⁵; and

Z is -O-, -S-, or -N(lower alkyl)-;

or a pharmaceutically acceptable salt thereof.

A preferred embodiment of the invention is a compound of Formula II wherein:

Y is

$$H_2NSO_2$$
 C_1 H_2NSO_2 H_2NSO_2 H_2NSO_2 C_1 H_2NSO_2 C_1 H_2NSO_2 C_1 C_1 C_1 C_1 C_1 C_1 C_1 C_1 C_2 C_1 C_2 C_1 C_1 C_2 C_1 C_1 C_1 C_2 C_1 C_1 C_1 C_2 C_1 C_2 C_1 C_1 C_2 C_1 C_1 C_2 C_1 C_2 C_1 C_1 C_2 C_1 C_1 C_2 C_1 C_2 C_1 C_1 C_2 C_2 C_1 C_1 C_2 C_2 C_1 C_2 C_1 C_2 C_2 C_2 C_1 C_2 C_2 C_2 C_1 C_2 C_2 C_2 C_1 C_2 C_2 C_2 C_2 C_2 C_1 C_2 C_2

Another preferred embodiment of the invention is a compound of Formula II wherein:
Y is

One embodiment of the invention is the use of the combinatorial library of Formula I in assays to discover biologically active compounds (ligands) of Formula II. Thus, an aspect of the invention is a method of identifying a compound having a desired characteristic which comprises synthesizing a combinatorial library of Formula I and testing the compounds of Formula I and the ligands of Formula II, either attached to the solid support or detached therefrom, in an assay which identifies compounds having the desired characteristic. A further embodiment of the invention is determining the structure of any compound so identified.

Another embodiment of the invention is an improved process for preparing t-butyl 4-(hydroxymethyl)-3-nitrobenzoate from t-butyl 4-(acetoxymethyl)-3-nitrobenzoate which comprises hydrazinolysis of the acetoxymethyl-nitrobenzoate using hydrazine or hydrazine hydrates in polar solvents such as lower alkanols, e.g.

methanol and isopropanol, and aprotic polar solvents such as acetonitrile at 0-50 °C. The improved process produces the hydroxymethyl-nitrobenzoate in nearly quantitative yield. When 4-(acetoxymethyl)-3-nitrobenzoic acid is linked to the solid support., the improved process for removing the alcohol protecting group, i.e., with hydrazine and methanol, is highly superior to the prior art methods.

Another embodiment of the invention is the use of divinylbenzene-cross-linked, polyethyleneglycol-grafted polystyrene beads optionally functionalized with amino groups (for example, TentaGel® S NH₂, Rapp Polymere) as the solid supports for constructing combinatorial libraries. Use of these beads greatly enhances the yield of ligands, e.g., compounds of Formula II, detached from the solid supports. Yields of ligands from supports such as DVB-crosslinked polystyrene are relatively low (<10%). The use of DVB-crosslinked, PEG-grafted polystyrene produces yields of >60%.

Definitions

The following abbreviations have the indicated meaning:

c- = cyclo

DEAD = diethylazodicarboxylate

20 DCM = dichloromethane = methylene chloride

DIC = diisopropylcarbodiimide

DMAP = 4-N,N-dimethylaminopyridine

DMF = N,N-dimethylformamide

DVB = 1,4-divinylbenzene

EDT = ethane dithiol

FACS = fluorescence activated cell sorting

Fmoc = 9-fluorenylmethoxycarbonyl

	GC	=	gas chromatography
	HOBt	-	N-hydroxybenzotriazole
	im	=	imidazole
	in	==	indole
5	m-	=	meta
	Me	=	methyl
	Mtr	=	4-methoxy-2,3,6-trimethylbenzenesulfonyl
	PEG	=	polyethylene glycol
	Ph	=	phenyl
10	r.t.	=	room temperature
	sat'd	=	saturated
	s-	= '	secondary
•	t-	=	tertiary
	TFA	=	trifluoroacetic acid
15	THF	=	tetrahydrofuran

Alkyl is intended to include linear, branched, or cyclic structures and combinations thereof. "Lower alkyl" includes alkyl groups of from 1 to 8 carbon atoms. Examples of lower alkyl groups include methyl, ethyl, propyl, isopropyl, butyl, s- and t-butyl, pentyl, hexyl, octyl, and the like. "Lower cycloalkyl" includes cycloalkyl groups of from 3 to 8 carbon atoms. Examples of lower cycloalkyl groups include c-propyl, c-butyl, c-pentyl, 2-methylcyclopropyl, cyclopropylmethyl, adamantly, and the like.

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"Alkenyl" is C3-C6 alkenyl of a linear, branched, or cyclic (C5-C6) configuration and combinations thereof. Examples of alkenyl

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groups include allyl, isopropenyl, pentenyl, hexenyl, c-hexenyl, 1-propenyl, 2-butenyl, 2-methyl-2-butenyl, and the like.

"Alkynyl" is C₃-C₆ alkynyl of a linear or branched configuration and combinations thereof. Examples of alkenyl groups include propyne, butyne, pentyne, 3-methyl-1-butyne, 3,3-dimethyl-1-butyne, and the like.

"Alkoxy" means alkoxy groups of from 1 to 6 carbon atoms of a straight, branched, or cyclic configuration. Examples of alkoxy groups include methoxy, ethoxy, propoxy, isopropoxy, cyclopropyloxy, cyclohexyloxy, and the like.

"Acylamino" means acylamino groups of from 1 to 6 carbon atoms of a straight, branched or cyclic configuration. Examples of acylamino groups are acetylamino, butylamino, cyclohexylamino, and the like.

Halogen includes F, Cl, Br, and I.

R² is intended to include racemates and all optical isomers. The amino acid residues of R² are, for example, methyl (alanine), hydroxymethyl (serine), phenylmethyl (phenylalanine), thiomethyl (cysteine), carboxyethyl (glutamic acid), etc.

In R⁴, "heteroaryl" means an aromatic 5- to 10-membered mono- or bicyclic heterocyclic ring system containing 1 or 2 N atoms, the systems either unsubstituted or substituted with 1-2 substituents selected from halogen, alkoxy, alkyl, CF₃, CN, -N(lower alkyl)₂, and acylamino;

In R⁵, the aromatic 6- to 10-membered carbocyclic rings include benzene and naphthalene and the 5- to 10- membered aromatic heterocyclic rings include imidazole, pyridine, and indole.

In A, the primary or secondary amino acids are intended to include alanine, asparagine, aspartic acid, arginine, cysteine, glutamic acid, glutamine, glycine, histidine, isoleucine, leucine, lysine, methionine, phenylalanine, proline, serine, threonine, tryptophan, tyrosine, valine, sarcosine, L-alanine, chloro-L-alanine hydrochloride, 2-aminoisobutyric acid, 2-(methylamino)isobutyric acid, DL-3-aminoisobutyric acid, DL-2-aminoisobutyric acid, (R)-(-)-2-

aminoisobutyric acid, (S)-(+)-2-aminoisobutyric acid, D-t-leucine, L-t-leucine, D-norvaline, L-norvaline, L-2-amino-4-pentenoic acid, D-

isoleucine, L-isoleucine, D-norleucine, 2,3-diaminopropionic acid monohydrochloride, L-norleucine, DL-2-aminocarprylic acid, β-alanine, DL-3-aminobutyric acid, 4-aminobutyric acid, 4-(methylamino)butyric acid hydrochloride, 5-aminovaleric acid, 7-aminoheptanoic acid, 8-aminocaprylic acid, 11-aminodecanoic acid, 12-aminododecanoic acid, carboxymethoxylamine hemihydrate, D-serine, D-homoserine, L-homoserine, L-(+)-canavaninesulfate homohydrate, D-allothreonine, L-allothreonine, D-threonine, L-threonine, DL-4-amino-3-hydroxybutyric acid, DL-3-

- hydroxynorvaline, (3S,4S)-(-)-statine, 5-hydroxy-DL-lysine hydrochloride, 5-aminoleucinic acid hydrochloride, 1-amino-1-cyclopropanecarboxylic acid, 1-amino-1-cyclopentanecarboxylic acid, 1-amino-1-cyclohexanecarboxylic acid, 5-amino-1,3-cyclohexadiene-1-carboxylic acid hydrochloride, 2-amino-2-norbornanecarboxylic acid,
- 15 (S)-(-)-2-azetidinecarboxylic acid, cis-4-hydroxy-D-proline, cis-4-hydroxy-L-proline, trans-4-hydroxy-L-proline, 3,4-dehydro-DL-proline, 3,4-dehydro-L-proline, D-pipecolinic acid, L-pipecolinic acid, mimosine, 2,3-diaminopropionic acid monohydrobromide, DL-2,4-diaminobutyric acid dihydrochloride, (S)-(+)-diaminobutyric acid
- 20 hydrochloride, D-ornithine hydrochloride, L-ornithine hydrochloride, 2-methylornithine hydrochloride monohydrate, N-ε-methyl-L-lysine hydrochloride, N-methyl-D-aspartic acid monohydrate, DL--2-methylglutamic acid hemihydrate, DL-2-aminoadipic acid, D-2-aminoadipic acid, L-2-aminoadipic acid, (+/-)-3-aminoadipic acid, D-
- cysteine hydrochloride monohydrate, D-penicillamine, Lpenicillamine, DL-homocysteine, S-methyl-L-cysteine, L-methionine,
 D-ethionine, L-ethionine, S-carboxymethyl-L-cysteine, (S)-(+)-2phenylglycine, (R)-(-)-2-phenylglycine, N-phenylglycine, N-(4hydroxyphenyl)glycine, D-phenylalanine, (S)-(-)indoline-2-carboxylic
- 30 acid, α-methyl,DL-phenylalanine, β-methyl-DL-phenylalanine hydrochloride, D-homophenylalanine, L-homophenylalanine, DL-2-fluorophenylglycine, DL-2-fluorophenylalanine, DL-3-fluorophenylalanine, DL-4-fluorophenylalanine, DL-4-chlorophenylalanine, 4-bromo-DL-
- phenylalanine, 4-iodo-D-phenylalanine, 3,3',5-triiodoL-thyronine, (+)-3,3',5-triiodo-L-thyronine sodium salt, D-thyronine, L-thyronine,

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DL-m-tyrosine, D-4-hydroxyphenylglycine, D-tyrosine, L-tyrosine, o-methyl-L-tyrosine, 3-fluoro-DL-tyrosine, 3-iodo-L-tyrosine, 3-nitro-L-tyrosine, 3,5-diiodo-L-tyrosine dihydrate, DL-dopa, L-dopa, 2,4,5-trihydroxyphenyl-DL-alanine, 3-amino-L-tyrosine

dihydrochloride monohydrate, 4-amino-D-phenylalanine hydrate, 4-amino-L-phenylalanine hydrate, 4-amino-DL-phenylalanine hydrate, 4-nitro-L-phenylalanine monohydrate, 4-nitro-DL-phenylalanine, 3,5-dinitro-L-tyrosine monohydrate, DL-α-methyltyrosine, L-α-methyltyrosine, (-)-3-(3,4-dihydroxyphenyl)-2-methyl-L-alanine sesquihydrate, DL-threo-3-phenylserine hydrate, and DL-DOPS.

Q is intended to include phenyl, thienyl, furanyl, thiadiazolyl, pyridyl, and pyrimidyl.

L and L' are depicted in Table 1, which also shows cleavage reagents. In designing a synthetic scheme, L and L' are chosen such that they are orthogonally reactive, i.e., they must allow for removal of either T or II (where T = T'-OH) without removal of the other since simultaneous cleavage of both T and II from the solid support is disadvantageous. In the structures as shown, the left-hand bond is the point of attachment to the solid support and the right-hand bond is the point of attachment to either T or II.

The tags of this invention, T, are chemical entities which possess several properties: they must be detachable from the solid supports, preferably by photolysis or oxidation; they must be individually differentiable, and preferably separable; they must be stable under the synthetic conditions; they must be capable of being detected at very low concentrations, e.g., 10-18 to 10-9 mole; they should be identifiable with readily-available equipment which does not require sophisticated technical capabilities to operate; and they should be relatively economical. The tags may be structurally related or unrelated, e.g., a homologous series, repetitive functional groups, related members of the Periodic Chart, different isotopes, combinations thereof, and the like. At the end of the combinatorial synthesis, to each solid support, there will usually be attached at least 0.01 femtomol. usually 0.001-50 pmol, of each tag. The tags may be aliphatic, alicyclic, aromatic, heterocyclic, or combinations thereof. Distinguishing features may be the number of repetitive units, such as methylene

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groups in an alkyl moiety; alkyleneoxy groups in a polyalkyleneoxy moiety; halo groups in a polyhalo compound; α - and/or β -substituted ethylene groups where the substituents may be alkyl groups, oxy, carboxy, amino, halo, or the like; isotopes; etc.

The materials upon which the combinatorial syntheses of the invention are performed are referred to as solid supports, beads, and resins. These terms are intended to include:

- a) beads, pellets, disks, fibers, gels, or particles such as cellulose beads, pore-glass beads, silica gels, polystyrene beads
 optionally cross-linked with divinylbenzene and optionally grafted with polyethylene glycol and optionally functionalized with amino, hydroxy, carboxy, or halo groups, grafted co-poly beads, poly-acrylamide beads, latex beads, dimethylacrylamide beads optionally cross-linked with N,N'-bis-acryloyl ethylene diamine, glass particles coated with
- hydrophobic polymer, etc., i.e., material having a rigid or semi-rigid surface; and
 - b) soluble supports such as low molecular weight non-cross-linked polystyrene.
- It is intended that the definitions of any substituent or symbol (e.g., R², R⁸, m, etc.) in a particular molecule be independent of its definitions elsewhere in the molecule.

TABLE 1 LINKER GROUPS

	LINK	ER GROUPS		
	Linker Group	Cleavage Reagent		
5	NO_2 $-CH_2B- \text{ or }$ NO_2 $-CH_2O - B-$	hν		
10	2. O ₂ N————————————————————————————————————	hv		
	3. OR O-	Ce(NH ₄) ₂ (NO ₃) ₆		
15	4. RO	Ce(NH ₄) ₂ (NO ₃) ₆		
	5. — $CH=CH(CH_2)_2$ -	O ₃ , OsO ₄ /IO ₄ -, or KMnO ₄		
	6. — CH=CHCH ₂ -	O ₃ , OsO ₄ /IO ₄ -, or KMnO ₄		
	7. —CH ₂ CH=CH-	O ₃ , OsO ₄ /IO ₄ -, or KMnO ₄		
20	8. 40 0-	1) O ₂ or Br ₂ , MeOH 2) H ₃ O ⁺		
	9. —CH=CHCH ₂ O-	(Ph ₃ P) ₃ RhCl(H)		
	10. Br	Li, Mg, or BuLi		
25	11. —S-CH ₂ -O-	Hg ⁺²		
	12. CH ₂ -O-	Zn or Mg		
	13. — CH ₂ -O-	Oxidation, e.g., Pb(OAc) ₄ or H ₅ IO ₆		
	D 77 1 11 11			

30 R = H or lower alkyl

X = electron withdrawing group such as Br, Cl, and I.

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Optical Isomers - Diastereomers - Geometric Isomers

Some of the compounds described herein contain one or more asymmetric centers and may thus give rise to enantiomers, diastereomers, and other stereoisometric forms which may be defined in terms of absolute stereochemistry as (R) or (S), or as (D) or (L) for amino acids. The present invention is meant to comprehend all such possible diastereomers as well as their racemic and optically pure forms. Optically active (R) and (S), or (D and L), isomers may be prepared using chiral synthons or chiral reagents, or resolved using conventional techniques. When the compounds described herein contain olefinic double bonds, and unless specified otherwise, it is intended to include both E and Z geometric isomers.

Salts

The pharmaceutical compositions of the present invention

comprise a compound of Formula II as an active ingredient or a
pharmaceutically acceptable salt thereof, and may also contain a
pharmaceutically acceptable carrier and, optionally, other therapeutic
ingredients. The term "pharmaceutically acceptable salts" refers to salts
prepared from pharmaceutically acceptable non-toxic acids or bases

including organic and inorganic acids or bases.

When a compound of the present invention is acidic, salts may be prepared from pharmaceutically acceptable non-toxic bases. Salts derived from all stable forms of inorganic bases include aluminum, ammonium, calcium, copper, iron, lithium, magnesium, manganese, potassium, sodium, zinc, etc. Particularly preferred are the ammonium, calcium, magnesium, potassium, and sodium salts. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary, and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines and basic ion-exchange resins such as arginine, betaine, caffeine, choline, N,N'-dibenzylethylenediamine, diethylamine, 2-diethylaminoethanol, 2-dimethylaminoethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, isopropylamine, lysine, methylglucosamine,

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morpholine, piperazine, piperidine, polyamine resins, procaine, purine, theobromine, triethylamine, trimethylamine, tripropylamine, etc.

When a compound of the present invention is basic, salts may be prepared from pharmaceutically acceptable non-toxic acids. Such acids include acetic, benzenesulfonic, benzoic, camphorsulfonic, citric, ethanesulfonic, fumaric, gluconic, glutamic, hydrobromic, hydrochloric, isethionic, lactic, maleic, mandelic, methanesulfonic, mucic, nitric, pamoic, pantothenic, phosphoric, succinic, sulfuric, tartaric, p-toluenesulfonic, etc. Particularly preferred are citric, hydrobromic, maleic, phosphoric, sulfuric, and tartaric acids.

In the discussion of methods of treatment herein, reference to the compounds of Formula II is meant to also include the pharmaceutically acceptable salts thereof.

Utilities

The ability of the compounds of Formula II to inhibit serine proteases makes them useful for preventing or reversing the symptoms induced by these enzymes in a mammal. This enzyme inhibition indicates that the compounds are useful to treat, prevent, or ameliorate hyper-coagulation disease in mammals, especially in humans, and are useful in post-myocardial infarction treatment and in post-angioplasty treatment.

The ability of compounds to inhibit metallo proteases such as angiotensin converting enzyme makes them useful for preventing or reversing the symptoms induced by these enzymes in a mammal. This enzyme inhibition indicates that the compounds are useful to treat, prevent, or ameliorate hypertension in mammals, especially in humans.

The ability of the compounds of Formula II to inhibit carbonic anhydrase isozymes makes them useful for preventing or reversing the symptoms induced by these enzymes in a mammal. This enzyme inhibition indicates that the compounds are useful to treat, prevent, or ameliorate ocular diseases, particularly glaucoma in mammals, especially in humans.

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Dose Ranges

The magnitude of the prophylactic or therapeutic dose of the compounds of Formula II will vary with the nature and severity of the condition to be treated and with the particular compound of Formula II and its route of administration. In general, the daily dose range for anti-enzymic use lies in the range of 20 to 0.001 mg/kg body weight of a mammal, preferably 10 to 0.01 mg/kg, and most preferably 1.0 to 0.1 mg/kg, in single or divided doses. In some cases, it may be necessary to use doses outside these ranges.

When a composition for intravenous administration is employed, a suitable daily dosage range is from about 10 to 0.0005 mg (preferably 5 to 0.01 mg) compound of Formula II per kg body weight.

When a composition for oral administration is employed, a suitable daily dosage range is from about 20 to 0.001 mg (preferably 10 to 0.01 mg) compound of Formula II per kg body weight.

When a composition for ophthalmic administration is employed, a suitable daily dosage range is from about 10-0.01% (preferably 5.0-0.5% compound of Formula II, typically prepared as a 2.0-0.1% by weight solution or suspension of a compound of Formula II in an acceptable ophthalmic formulation.

The compounds of Formula II may also be used in combination with other pharmaceutically active ingredients. For example, a typical ocular formulation may comprise the compound alone or in combination with a β -adrenergic blocking agent such as timolol maleate or a parasympathomimetic agent such as pilocarpine. When used in combination, the two active ingredients are present in approximately equal parts.

Pharmaceutical Compositions

Any suitable route of administration may be employed for providing a mammal, especially a human, with an effective dosage of a compound of Formula II. For example, oral, rectal, topical, parenteral, ocular, pulmonary, nasal, etc. routes may be employed. Dosage forms

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include tablets, troches, dispersions, suspensions, solutions, capsules, creams, ointments, aerosols, and the like.

The pharmaceutical compositions of the present invention comprise a compound of Formula II, or a pharmaceutically acceptable salt thereof, as an active ingredient, and may also contain a pharmaceutically acceptable carrier and, optionally, other therapeutically active ingredients.

The compositions include compositions suitable for oral, rectal, topical (including transdermal devices, aerosols, creams, ointments, lotions, and dusting powders), parenteral (including 10 subcutaneous, intramuscular, and intravenous), ocular (ophthalmic), pulmonary (nasal or buccal inhalation), or nasal administration; although the most suitable route in any given case will depend largely on the nature and severity of the condition being treated and on the nature of the active ingredient. They may be conveniently presented in unit dosage form and prepared my any of the methods well known in the art of pharmacy.

A compound of Formula II may be combined as the active ingredient in intimate admixture with a pharmaceutical carrier according to conventional pharmaceutical compounding techniques. The carrier may take a wide variety of forms depending on the nature of the preparation desired for administration, i.e., oral, parenteral, etc. In preparing oral dosage forms, any of the usual pharmaceutical media may be used, such as water, glycols, oils, alcohols, flavoring agents, preservatives, coloring agents, and the like in the case of oral liquid preparations (e.g., suspensions, elixirs, and solutions); or carriers such as starches, sugars, microcrystalline cellulose, diluents, granulating agents, lubricants, binders, disintegrating agents, etc. in the case of oral solid preparations such as powders, capsules, and tablets. Solid oral preparations are preferred over liquid oral preparations. Because of their ease of administration, tablets and capsules are the preferred oral dosage unit form. If desired, capsules may be coated by standard aqueous or non-aqueous techniques.

In addition to the dosage forms described above, the compounds of Formula II may be administered by controlled release means and devices such as those described in U.S.P. Nos. 3,536,809; 3,598,123; 3,630,200; 3,845,770; 3,916,899; and 4,008,719, which are incorporated herein by reference.

Pharmaceutical compositions of the present invention suitable for oral administration may be prepared as discrete units such as capsules, cachets, or tablets each containing a predetermined amount of the active ingredient in powder or granular form or as a solution or suspension in an aqueous or nonaqueous liquid or in an oil-in-water or 10 water-in-oil emulsion. Such compositions may be prepared by any of the methods known in the art of pharmacy. In general, the compositions are prepared by uniformly and intimately admixing the active ingredient with liquid carriers, finely divided solid carriers, or both and then, if necessary, shaping the product into the desired form. For 15 example, a tablet may be prepared by compression or molding, optionally with one or more accessory ingredients. Compressed tablets may be prepared by compressing in a suitable machine the active ingredient in a free-flowing form such as powder or granule optionally mixed with a binder, lubricant, inert diluent, or surface active or 20 dispersing agent. Molded tablets may be made by molding in a suitable machine, a mixture of the powdered compound moistened with an inert liquid diluent. Opthalmic inserts are made from compression molded films which are prepared on a Carver Press by subjecting the powdered mixture of active ingredient and HPC to a compression force of 12,000 25 lb.(gauge) at 149 °C for 1-4 min. The film is cooled under pressure by having cold water circulate in the platen. The inserts are then individually cut from the film with a rod-shaped punch. Each insert is placed in a vial, which is then placed in a humidity cabinet (88% relative humidity at 30 °C) for 2-4 days. After removal from the 30 cabinet, the vials are capped and then autoclaved at 121 °C for .0.5 hr.

The following are representative pharmaceutical dosage forms of the compounds of Formula II:

	I.M. Injectable Suspension	mg/mL
	Compound of Formula II	10
	Methylcellulose	5
	Tween 80	0.5
5	Benzyl alcohol	9
	Benzalkonium chloride	1
	Water for injection to a total volume of 1 mL	
*	<u>Tablet</u>	mg/tablet
	Compound of Formula II	25
10	Microcrystalline cellulose	415
	Povidone	14
	Pregelatinized starch	43.5
	Magnesium stearate	<u>2.5</u>
		500
15	Capsule	mg/capsule
	Compound of Formula II	25
	Lactose powder	573.5
	Magnesium stearate	<u>1.5</u>
		600
20	Aerosol	Per canister
	Compound of Formula II	24 mg
	Lecithin, NF liquid concentrate	1.2 mg
	Trichlorofluoromethane, NF	4.025 gm
	Dichlorodifluoromethane, NF	12.15 gm

	Opthalmic Solution	mg/mL
	Compound of Formula II	1
	Monobasic sodium phosphate•2H ₂ O	9.38
	Dibasic sodium phosphate-12H2O	28.48
5	Benzalkonium chloride	1
	Water for injection to a total volume of 1 mL	•
	Opthalmic Suspension	mg/g
	Compound of Formula II	1
	Petrolatum liquid to a total weight of 1 g	-
10	Opthalmic Insert	mg/insert
	Compound of Formula II	1
	Hydroxypropylcellulose	12
	• • • •	14

These compounds may also be used as libraries for discovering new lead structures by evaluation across an array of biological assays, including the discovery of selective inhibition patterns 15 across isozymes. These libraries are thus tools for drug discovery; i.e., as a means to discover novel lead compounds by screening the libraries against a variety of biological targets and to develop structure-activity relationships in large families of related compounds. The libraries may be tested with the ligands attached to the solid supports as depicted in 20 Formula I or the individual compounds II may be detached prior to evaluation. With the compounds of Formula I, screening assays such as FACS sorting and cell lawn assays may be used. When a compound is detached prior to evaluation, its relationship to its solid support is maintained, for example, by location within the grid of a standard 96-25 well plate or by location of activity on a lawn of cells. The solid support associated with bioactivity or the solid support related to the detached ligand may then be decoded to reveal the structural or synthetic history of the active compound (Ohlmeyer et al., Proc. Natl.

Acad. Sci. USA, 90, 10922-10926, Dec. 1993).

used:

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Assays for Determining Biological Activity

Xanthine Oxidase Inhibition - The following materials are

3.9 µM hypoxanthine

0.3 mM 4-aminoantipyrene

2 mM 3,5-dichloro-2-hydroxybenzenesulfonate

50 mM sodium phosphate buffer, pH 7.5

5 U/mL horseradish peroxidase (Sigma P-6782, 5500 U/5 mg)

3 nM xanthine oxidase (buttermilk, Sigma X-4500, 16 U/mL)

10 inhibitor

Reactions are carried out in 24 μL total volume in 96-well U-bottom polypropylene microtiter dishes (Costar) containing the test compounds. 8 μL of sodium phosphate buffer, pH 7.5, is added to each well. A substrate mixture is prepared on ice by mixing 0.53 mL sodium phosphate buffer, 0.4 mL 4-aminoantipyrene (0.61 mg/mL), 0.4 mL 3,5-dichloro-2-hydroxybenzene-sulfonate (5.3 mg/mL), 4 μL horseradish peroxidase (Sigma P-6782, 5500 U/5 mg), and 128 μL hypoxanthine 920 μg/mL. 8 μL of the substrate mixture is then pipetted into each well. 8 μL xanthine oxidase (buttermilk, 9.0 nM, Sigma X-4500, 16 U/mL) in sodium phosphate buffer, pH 7.5 (or buffer alone as a control) is added last, directly into the reaction mixture. The plates are pulse-spun briefly in a tabletop centrifuge before reading absorbance. Absorbance is read using a dual kinetics program (490 minus 650 nm) for 15 min. at r.t. without automix, in a microplate

Other assays for evaluating the compounds of the present invention are well known in the art. Representative examples of references teaching these assays follow:

reader (Molecular Devices Thermomax). Initial rates are calculated

(Vmax program) and compared to those of reactions without inhibitor.

30 <u>ACE Inhibition</u> - Holmquist <u>et al.</u>, "A Continuous Spectrophotometric Assay for Angiotensin Converting Enzyme", *Anal. Biochem.*, <u>95</u>, 540-548 (1979).

<u>Thrombin Inhibition</u> - Lottenberg <u>et al.</u>, "Assay of Coagulation Proteases Using Peptide Chromogenic and Fluorogenic Substrates", *Meth. in Enzymol.*, <u>80</u>, 341-361, (1981).

Carbonic Anhydrase Inhibition - Maren and Couto, "The Nature of Anion Inhibition of Human Red Cell Carbonic Anhydrases", Archiv. of Biochem. and Biophy., 196, No. 2, Sept., 501-510 (1979).

<u>Carbonic Anhydrase Inhibition</u> - Ponticello <u>et al.</u>, "Thienothiopyran-2-sulfonamides: A Novel Class of Water-Soluble Carbonic Anhydrase Inhibitors", *J. Med. Chem.*, <u>30</u>, 591597 (11987).

10 Methods of Synthesis

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The compounds of the present invention can be prepared according to the following methods. At each step in the synthesis each solid support upon which a compound is being synthesized is uniquely tagged to define the particular chemical event(s) occurring during that step. The tagging is accomplished using identifiers such as those of Formula IV, which record the sequential events to which the support is exposed during the synthesis, thus providing a reaction history for the compound produced on each support. The identifiers are used in combination with one another to form a binary or higher order encoding scheme permitting a relatively small number of identifiers to encode a relatively large number of reaction products. For example, when used in a binary code, N identifiers can encode up to 2N different compounds and/or conditions. By associating each variable or combination of variables at each step of the synthesis with a combination of identifiers which uniquely define the chosen variables such as reactant, reagent, reaction conditions, or combinations of these, one can use the identifiers to define the reaction history of each solid support.

In carrying out the syntheses, one begins with at least 10³, desirably at least 10⁴, and generally not exceeding 10¹⁵ solid supports.

Depending on the pre-determined number of R¹ choices for the first step, one divides the supports accordingly into as many containers. The appropriate reagents and reaction conditions are applied to each container and the combination of identifiers which encode for each R¹ choice is added and attached. Depending on the chemistries involved,

the tagging may be done prior to, concomitantly with, or after the reactions which comprise each choice. As a control, sample supports may be picked at any stage and a portion of their tags detached and decoded to verify that the correct tags are bound to the sample supports.

As needed, one may wash the beads free of any excess reagents or byproducts before proceeding. At the end of each step, the supports are
combined, mixed, and again divided, this time into as many containers
as pre determined for the number of A choices for the second step in
the synthesis. This procedure of dividing, reacting, tagging, and
remixing is repeated until the combinatorial synthesis is completed.

Scheme 1

washed in CH₂Cl₂

Functionalized supports such as amino-functionalized or hydroxy-terminating PEG grafted polystyrene beads are equally divided into a pre-determined number of reaction vessels and are reacted with either a) a cleavable linker/ligand element which has been pre-formed to 15 generate 2 when the detached ligand element terminates in OH or b) a cleavable linker, followed by reaction with a ligand element to generate 3 in the case when the detached ligand element terminates in COOH. Compounds $\underline{2}$ and $\underline{3}$ are then treated with piperidine/DMF to de-protect the amino group of the ligand element R¹ to yield 4 (Scheme 1, step 20 (c)). Unique tagging of the supports in each reaction vessel is achieved with combinations of identifiers encoded in a binary scheme, e.g., as depicted in Table 1-1 for seven choices of R1. The identifiers are attached by adding a solution of the identifiers (in a 5% wt./wt. identifier:solid support ratio) to a batch of supports suspended in 25 CH₂Cl₂ and shaking the mixture for 30 min. A dilute solution of rhodium trifluoroacetate dimer is added and the mixture is immediately shaken. The mixture is continued to be shaken overnight and the

Scheme 2

The compounds 4 are pooled, mixed, and then divided into a pre-determined number of reaction vessels, each of which is treated with one reagent corresponding to ligand element A, in the presence of HOBt/DMF to produce 5. The reaction vessels are then treated with piperidine/DMF to de-protect the terminal amino group, yielding 6. Unique tagging of the supports in each reaction vessel is achieved with combinations of additional identifiers encoded in a binary scheme, e.g., as depicted in Table 1-2 for 31 choices of A.

10 <u>Scheme 3</u>

The compounds 6 are pooled, mixed, and then divided into a pre-determined number of reaction vessels each of which is treated with one reagent corresponding to ligand element Y in the presence of solvents such as CH₂Cl₂ and DMF, and, when required, condensation reagents such as DIC, acylation catalysts such as DMAP, and bases such 15 as triethylamine to produce 7. Unique tagging of the supports in each reaction vessel is achieved with combinations of additional identifiers encoded in a binary scheme, e.g., as in Table 1-3 for 31 choices of Y. Compound 7 is then either exposed to UV light (~360 nm) in a lower alkanol such as MeOH to cleave the protected form of the compounds of 20 Formula II from the support/linker complex or first treated with TFA/thioanisole/EDT to remove the protecting groups on the R² sidechains and then exposed to UV light in a lower alkanol such as MeOH to cleave compound II.

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SCHEME 1 ADDITION OF R¹

a. Carbonate-linked Moieties

S NH₂

Reagents; e.g. 1-4,
DIC, DMAP

methylenechloride

S H
NO₂
O
R¹-Fmoc

10 <u>2</u>

$$R^1 =$$

20 Reagents 1-4

$$R^{1} = 0$$

$$1$$

$$R^{1} = \sqrt{0}$$

$$2$$

$$R^{1} = \bigvee_{N = 1}^{N} R^{1} = \bigvee_{N = 1}^{N} A$$

SCHEME 1 (Cont.)

b. Ester-linked Moieties

c. Fmoc Deprotection

SCHEME 2 ADDITION OF A

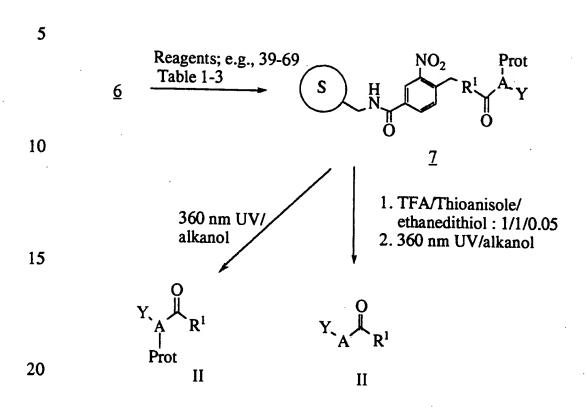
15
$$\underline{5}$$
 $\underline{50\% \text{ piperidine/DMF}}$ $\underline{5}$ \underline{H} $\underline{NO_2}$ $\underline{R^1}$ \underline{A} \underline{H} \underline{O}

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(Prot) = optional, non-Fmoc, base stable sidechain protecting group

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SCHEME 3 ADDITION OF Y



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Representative Compounds

Table 2 illustrates compounds of Formula II which are representative of the present invention:

Y-A-CO-R¹

TABLE 2

REPRESENTATIVE COMPOUNDS

II

10			
10	<u>Y</u>	A	R ¹
15	1. H ₂ NO ₂ S S	-NH(CH ₂) ₂ -	-NH(CH ₂) ₂ O(CH ₂) ₄ OH
13	2. H ₂ NO ₂ S S	-NH(CH ₂) ₃ -	-NH(CH ₂) ₃ COOH
20	3. H ₂ NO ₂ S O	-NH(CH ₂) ₅ -	-NH(CH ₂) ₄ OH
	4. H ₂ NO ₂ S / S	-NH(CH ₂) ₁₀ -	_N(CH ₂) ₂ -ОН
25	5. H	-N_	−N (CH ₂) ₃ -OH

TABLE 2 (Cont.)

	O N H 6. N	-N	-N(CH ₂) ₃ -OH
5	CI CI—SO ₂ -	CH₃ -NHCH-	-N(CH ₂) ₄ -CO ₂ H
10	F ₃ C	CH₂OH -NHCH-	CH ₃ -N-(CH ₂) ₄ -OH
15	H_2NSO_2 —CO-	H ₂ C-CH(CH ₃) ₂ -NHCH-	−N −CH ₂ OH
20	CI CO- SO ₂ NH ₂	(CH ₂) ₂ COOH -NHCH- (L)	−N CH ₂ OH

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Table 3 illustrates additional compounds of Formula II representative of the present invention which are inhibitors of carbonic anhydrase:

Y-A-CO-R¹ II

PCT/US95/03223

TABLE 3
REPRESENTATIVE COMPOUNDS

Y \mathbb{R}^1 CO-·CH(CH₃)₂ 10 -NHCH-СООН (L) 1. H₂NSO₂ .CO--NHCH-15 CH₂OH 2. (L) H₂NSO₂ CO-CI. (CH₂)₂-COOH CH₂OH -NH-CH-3. 20 (L) SO₂NH₂ CO- .. -NH-(CH₂)₅-CI 4. OH CI SO₂NH₂ 25 -NH-(CH₂)₅-COOH -NHCH-5. (L) H₂NSO 30

Table 3 (Cont.)

5 .	6. H ₂ NSO ₂ CO-	-NHCH- (L)	$-N$ $-CO_2H$
	CI CO- 7. SO ₂ NH ₂	-NH CH- (L)	$-N$ $-CO_2H$
10	8. CI SO ₂ NH ₂	CH₃ -NHCH- (D)	-NH-(CH ₂) ₅ -COOH
13	9. H ₂ NSO ₂ CO-	-NHCH- (D)	-NCO₂H
20	10. CI SO ₂ NH ₂	CH₃ -NH·CH- (D)	-N_OH
25	11. H ₂ NSO ₂ CO-	-NHCH- (D)	-N_OH

Table 3 (Cont.)

5	12. H ₂ NSO ₂ CO-	-N-(L)	-N_СООН
•	13. H ₂ NSO ₂ CO-	CH(CH ₃) ₂ -NH CH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
10	14. H ₂ NSO ₂ CO-	(CH ₂) ₂ -COOH -NH-CH- (L)	-NH-(CH ₂) ₅ -COOH
15	CI CO- 15. CI SO ₂ NH ₂	CH ₂ C(O)NH ₂ -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
20	16. H ₂ NSO ₂	CH ₂ CO ₂ H -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
25	17. H ₂ NSO ₂ CO-	CH ₂ CO ₂ H -NHCH- (D)	−N—CO ₂ H
30	18. H ₂ NSO ₂ CO-	CH(CH ₃) ₂ -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H

Table 3 (Cont.)

5	19. H ₂ NSO ₂ CO-	CH ₂ CH ₂ SCH ₃ -NHCH- (D)	-N_OH
·	20. H ₂ NSO ₂	-NH(CH ₂) ₆ -	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
10	21. H ₂ NSO ₂	H ₃ C-CHCH ₂ CH ₃ \ -NHCH- (L)	-мон
15	22. H ₂ NSO ₂ CO-	(CH ₂) ₂ CO ₂ H -NHCH- (D)	−N—CO ₂ H
20	23. H ₂ NSO ₂	CH(CH ₃) ₂ -NHCH- (D)	-ион
25	24. H ₂ NSO ₂ CO-	CH ₂ CH(CH ₃) ₂ -NHCH- (L)	-ион
	25. H ₂ NSO ₂ CO-	-N-	-Nон
30			

Table 3 (Cont.)

5	26. H ₂ NSO ₂	CH ₂ CH(CH ₃) ₂ -NHCH- (L)	-NH-(CH ₂) ₅ -COOH
	27. H ₂ NSO ₂ CO-	-NHCH- N (D)	−N CH ₂ OH
10	28. H ₂ NSO ₂ CO-	-Ph -NHCH- (L)	−N CH ₂ OH
15	29. H ₂ NSO ₂ CO-	COOH -NHCH- (L)	-м соон
20	30. H ₂ NSO ₂	C(O)NH ₂ -NHCH- (L)	-NH-(CH ₂) ₅ -COOH
25	31. H ₂ NSO ₂ CO-	-NH(CH ₂) ₅ -	-NCO₂H
30	32. H ₂ NSO ₂	CH₃ -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H

Table 3 (Cont.)

5	33. H ₂ NSO ₂	(CH ₂) ₂ CO ₂ H -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
10	34. H ₂ NSO ₂ CO-	(CH ₂) ₂ C(O)NH ₂ -NHCH- (D)	-N—CH₂OH
10	35. H ₂ NSO ₂ CO-	(CH ₂) ₂ C(O)NH ₂ -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
15	36. H ₂ NSO ₂	(CH ₂) ₂ CO ₂ H -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH Н

The invention is further defined by reference to the following examples, which are intended to be illustrative and not limiting.

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PREPARATION 1 IDENTIFIERS

Thirteen compounds of the general formula:

$$\begin{array}{c|c}
N_2 & O & (CH_2)_n - O - Ar \\
O & O & IV
\end{array}$$

wherein:

n = 3-12 and Ar is pentachlorophenyl or

n = 4-6 and Ar is 2,4,6-trichlorophenyl were prepared according to Scheme 4 and the following illustrative example.

- a) Methyl vanillate (0.729 g, 4.0 mmole), 1-hydroxy-95 (2,3,4,5,6-pentachlorophenoxy)nonane (1.634 g, 4.0 mmole) and triphenylphosphine (1,258 g, 4.8 mmole) were dissolved in 20 mL dry toluene under argon. DEAD (0.76 mL, 0.836 g, 4.8 mmole) was added dropwise and the mixture was stirred at 25°C for one hr. The solution was concentrated to half volume and purified by flash chromatography eluting with DMC to give 1.0 g (1.7 mmole, 43%) of the product as a white crystalline solid.
 - b) The methyl ester from Step (a) (1.0 g, 1.7 mmole) was dissolved in 50 mL THF, 2 mL water was added, followed by LiOH (1.2 g, 50 mmole). The mixture was stirred at 25°C for one hr. then refluxed for 5 hr. After cooling to 25°C, the mixture was poured onto ethyl acetate (200 mL) and the solution was washed with 1 M HCl (3x 50 mL) then sat'd aq. NaCl (1x 50 mL) and dried over sodium sulfate. The solvent was removed and the crude acid azeotroped once with toluene.
- c) The crude material from Step (b) was dissolved in 100 20 -mL toluene, 10 mL (1.63 g, 14 mmole) thionyl chloride was added, and the mixture was refluxed for 90 min. The volume of the solution was reduced to approx. 30 mL by distillation, then the remaining toluene was removed by evaporation. The crude acid chloride was dissolved in 20 mL dry DCM and cooled to -70°C under argon and a 25 solution of approx. 10 mmole diazomethane in 50 mL anhydrous ether was added. The mixture was warmed to r.t. and stirred for 90 min. Argon was bubbled through the solution for 10 min., then the solvents were removed by evaporation and the crude material was purified by flash chromatography, eluting with 10-20% ethyl acetate in hexane. 30 The diazoketone (0.85 g, 1.4 mmole, 82% yield over three steps) was obtained as a pale yellow solid.

An improvement was made to the final diazomethylation step, whereby the acid chloride was reacted with (trimethylsilyl)diazomethane and triethylamine to give the identifier,

which was then used without further purification. This was a significant improvement over the original reaction with diazomethane, as the identifier was now obtained in high yield with no chlorometylketone byproduct. Also, purification by flash chromatography was no longer necessary, which in some cases had resulted in significant acid-catalyzed decomposition of the identifier.

Alternate Step c) To a solution of the acyl chloride (3.8 mmol, 1.00 eq.) and 1.85 mL (13.3 mmol, 3.50 eq.) of triethylamine in anhydrous THF/acetonitrile (1:1) at 0°C under argon was added 5.7 mL (11.4 mmol, 3.00 eq.) of a 2.0 M solution of (trimethylsilyl)diazomethane in hexanes. The resulting orange solution was stirred at 0°C for 2 hr, then at 25°C for 17 hr. (If a precipitate formed immediately upon addition of (trimethylsilyl)diazomethane, CH₂Cl₂ was added until the precipitate redissolved). EtOAc was added (250 mL), and the organic layer washed with saturated aq. NaHCO₃ (100 mL) and H₂O (100 mL), then dried (anhydrous MgSO₄). Removal of the volatiles in vacuo gave the product as yellow crystals in 60-100% yield.

The other 12 identifiers of Formula IV were prepared by analogous synthetic routes, steps (a), (b), and (c).

In the synthesis of Example 1, the 13 identifiers were used to encode the combinatorial library. In Step 1, pentachlorophenyl identifiers where n = 10-12 (abbreviated C₁₀Cl₅, C₁₁Cl₅, and C₁₂Cl₅) were used in the following binary encoding scheme: 001 = (n = 12), 010 = (n = 11) and 100 = (n = 10). In Step 2, pentachlorophenyl identifiers where n = 5-9 (abbreviated C₅Cl₅, C₆Cl₅, C₇Cl₅, C₈Cl₅, and C₉Cl₅) were used and encoded as follows: 00001 = (n = 9), 00010 = (n = 8), 00100 = (n = 7), 01000 = (n = 6) and 10000 = (n = 5). In Step 3, pentachlorophenyl identifiers where n = 3-4 (abbreviated C₃Cl₅ and C₄Cl₅) were used and encoded as follows: 00001 = (n = 4), 00010 = (n = 3). Also in Step 3, trichlorophenyl identifiers where n = 4-6 (abbreviated C₄Cl₃,C₅Cl₃, and C₆Cl₃) were used and encoded as follows: 00100 = (n = 6), 01000 = (n = 5), and 10000 = (n = 4).

Thus, in Step 1 reagent 3 (Table 1-1) is encoded "011" which represents tagging this choice in the synthesis with the two pentachloro-phenyl identifiers where n = 11 and 12. Likewise, in Step 3 reagent 52 (Table 1-3) is encoded "01110" which represents tagging this choice in the synthesis with the pentachlorophenyl identifier where n = 3 and the two trichlorophenyl identifiers where n = 5 and 6.

SCHEME 4 IDENTIFIERS

PREPARATION 2 t-BUTYL-4-(HYDROXYMETHYL)-3-NITROBENZOATE

t-Butyl 4-(acetoxymethyl)-3-nitrobenzoate was prepared as described by Barany and Albericio, J. Am. Chem. Soc. 1985, 107, 4936-4942. However, attempts to also follow the reference's final procedure for hydrazinolysis of the acetate using hydrazine hydrate in CHCl3 at 25°C produced only trace amounts of the desired hydroxymethyl final product, which is the t-butyl ester pre-cursor of the photocleavable linker used herein. It has now been found that 10 hydrazinolysis using hydrazine hydrate in MeOH or DMF at 25 °C produces t-butyl 4-(hydroxymethyl)-3-nitrobenzoate in high yield but that a byproduct produced in DMF by additional diimide reduction of the nitro functionality to an amino group makes this solvent undesirable. 15 Using MeOH as solvent, only the desired final product was obtained in

near quantitative yield.

This new hydrazinolysis procedure was also used to produce the photocleavable linker when attached to TentaGel S NH2 resin.

- 20 t-Butyl 4-(hydroxymethyl)-3-nitrobenzoate: To a solution of 14.1 g (47.7 mmol, 1.00 eq.) of t-butyl 4-(acetoxymethyl)-3-nitrobenzoate in MeOH (200 mL) was added 27.0 mL (477 mmol, 10.0 eq.) of hydrazine hydrate (55% hydrazine). The resulting yellow solution was stirred at 25°C for 4 hr. EtOAc (250 mL) and saturated aq. NaCl (85 mL) were
- added, and the organic layer collected after shaking. The organic layer 25 was washed further with saturated aq. NaCl (2 x 85 mL), and then dried (MgSO₄). Removal of volatiles in vacuo gave the product in 93% yield as yellow crystals.
- TentaGel S NH₂ attached 4-(hydroxymethyl)-3-nitrobenzamide: See 30 Example 1, Step 1a.

The reaction is represented in Scheme 5.

SCHEME 5 HYDRAZINOLYSIS

$$(CH_{3})_{3}C-O-\overset{O}{C}-\overset{O}{\longleftarrow}CH_{2}\cdot O \overset{O}{\longrightarrow}CH_{3}$$

$$NO_{2}$$

$$MeOH, H_{2}NNH_{2} \cdot xH_{2}O \qquad (CH_{3})_{3}C-O-\overset{O}{\longleftarrow}CH_{2}OH$$

$$25^{\circ}C, 4 \text{ hr.} \qquad (98\% \text{ yield}) \qquad NO_{2}$$

EXAMPLE 1

6727 COMPOUND LIBRARY

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Step 1

a) Preparation of ester-linked resin batches

TentaGel S NH2, (10 g, 2.9 mmole, 0.29 mmole/g) was suspended in 60 mL methylenechloride. 4-acetoxymethyl-3-nitrobenzoic acid (2.77 g, 11.6 mmole, 4 eq), DMAP (1.42 g, 11.6 mmole, 4 eq) and DIC (1.81 mL, 1.46 g, 11.6 mmole, 4 eq) was added in that order. The mixture was agitated at 25 °C with a wrist action shaker for 5 hours, at which time the resin gave a negative Kaiser test. The derivatized resin was washed (methylenechloride 5 x 50 mL, isopropanol 5 x 50 mL, then methylenechloride 5 x 50 mL), dried in vacuo and stored in the dark.

The acetoxymethyl resin above was washed with methanol (1 x 150 mL), then suspended in 150 mL 10% hydrazine/methanol. The resin was agitated at 25 °C for 44 hours, filtered, then washed (methanol, $5 \times 100 \text{ mL}$), then methylene chloride, $5 \times 100 \text{ mL}$) and dried in vacuo.

Three batches of the above hydroxymethyl resin (3 g each, 0.84 mmole, 0.28 mmole/g) were placed in 100 mL synthesis vessels and each was suspended in 60 mL dichloromethane. One of N-Fmoc isonipecotic acid (0.885 g, 2.52 mmole, 3 eq), N-Fmoc nipecotic acid

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(0.885 g, 2.52 mmole, 3 eq), and N-Fmoc-ε-amino caproic acid (0.890 g, 2.52 mmole, 3 eq) was added to each of the three batches. DMAP (0.031 g, 0.252 mmole, 0.3 eq) and then DIC (0.4 mL, 0.318 g, 2.52 mmole, 3 eq) was added to each and the resin batches were agitated at 25 °C for 22 hours. The resin batches were filtered, washed (methylenechloride (5 x 50 mL) and isopropanol (2 x 50 mL), then methylenechloride (5 x 50 mL)), and dried in vacuo.

b) Preparation of carbonate-linked resin batches

N-Fmoc-2-hydroxymethylpiperidine (1.35 g, 4 mmole, 1.eq) was dissolved in 20 mL methylenechloride and cooled to 0 °C. Phosgene (8 mL, 2 M solution in toluene, 16 mmole, 4 eq) was added followed by 2,6-lutidine (1.86 mL, 1.71 g, 4 eq), then the mixture was stirred for 30 min. The mixture was concentrated on a rotary evaporator to give a viscous slurry and this residue was redissolved in 30 mL

- methylenechloride (the solution contained some undissolved solid). tButyl 4-hydroxymethyl-3-nitrobenzoate (0.51 g, 2 mmole, 0.5 eq) was added and the mixture was stirred at 25 °C for 2 hours. The crude reaction mixture was poured onto 200 mL ethyl acetate and this solution was washed with 1 M HCl (2 x 100 mL), sat'd aq. NaHCO3 (2 x 100 mL).
- mL) and sat'd aq. NaCl (1 x 100 mL). The solution was dried over magnesium sulfate, filtered, and evaporated to give the crude product. This was purified by flash chromatography eluting with 10-30% ethylacetate/hexane to give the product (1.22 g, 1.97 mmole, 99%) as a pale yellow solid.

The carbonate above was dissolved in 20 mL methylene chloride, 10 mL TFA was added and the mixture was stirred at 25 °C for 16 hours. The solution was diluted with 50 mL toluene and evaporated to give the carboxylic acid as a viscous yellow oil which was azeotroped with toluene once, then dried in vacuo.

The acid prepared above (~2 mmole, 2 eq) was discolved in 20.

The acid prepared above (~2 mmole, 2 eq) was dissolved in 30 mL methylenechloride and this solution was added to TentaGel S NH₂ (3 g, 0.32 mmole/g, ~1 mmole, 1 eq). The resin was suspended by agitation then DMAP (40 mg, 0.3 mmole, 0.3 eq) and DIC (0.47 mL, 0.4 g, 3 eq) were added in that order. The resin was agitated at 25 °C for 19 hours, at which time it gave a negative Kaiser test. The resin

was filtered and washed (methylenechloride 10 x 50 mL), then dried in vacuo.

The three other carbonate linked resin batches were prepared in an analogous manner using the reagents of Table 1-1.

5 c) Encoding of Step 1

One gram batches of the seven resin batches from Steps 1(a) and 1(b) with the appropriately linked N-Fmoc protected amino acid or amino alcohol were placed in seven separate synthesis vessels and each was suspended in 20 mL methylene chloride.

Stock solutions of 200 mg of C12Cl5, C11Cl5, and C10Cl5
-linker-diazoketone (Preparation 1) were prepared in 4 mL
methylenechloride. Aliquots (1 mL) of the stock solutions were used to
prepare the appropriate binary coding mixtures for each of the seven
resin batches. The appropriate coding mixture was added to each resin
batch and the resin was agitated for 1 hour. Rhodium trifluoroacetate
dimer (1 mL of a 1 mg/mL solution in methylenechloride) was added to
each of the vessels and the resin was agitated at 25 °C overnight. Each
resin batch was washed with methylenechloride (1 x 50 mL), then the
batches were combined and the entire library (seven compounds) was
washed with methylenechloride (10 x 50 mL).

d) Deprotection

N-Fmoc protecting groups were removed by washing the resin once with DMF, filtering, then suspending the resin in 60 mL 50% piperidine/DMF and agitating at room temperature for 30 min. The resin was filtered, washed (DMF (5 x 50 mL)) and methylenechloride (10 x 50 mL)) and dried in vacuo.

Step 2

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a) Addition of A

The dried resin from Step 1(d) was divided into 31 batches of 210 mg (~0.07 mmole), each of which was placed in a 20 mL synthesis vessel. Each of the reagents (0.25 g, > 0.4 mmole, > 6 eq) (Table 1-2) used in the second step of the synthesis (N-Fmoc amino acid with acid labile side chain protection where appropriate) was dissolved in 10 mL methylenechloride. HOBt (1 mL of 1 mg/mL in DMF) was added and

the solutions were shaken briefly. Further DMF was added to those amino acids that had not completely dissolved. Each reagent solution was added to one of the 31 resin batches. DIC (0.2 mL, ~1 mmole) was added to each vessel and the resin was agitated at 25 °C overnight.

Each of the resin batches was filtered and washed separately (DMF (1x 15 mL) and methylenechloride (10 x 15 mL)). The resin was suspended in 10 mL methylenechloride.

b) Encoding of Step 2

Stock solutions of 160 mg of C9Cl5, C8Cl5, C7Cl5, C6Cl5, and C5Cl5 -linker-diazoketone (Preparation 1) were prepared in 16 mL methylenechloride. Aliquots (1 mL) of the stock solutions were used to prepare the appropriate binary coding mixtures for each of the 31 resin batches (Table 1-2). The appropriate coding mixture was added to each of the resin batches and the resin was agitated for 30 min. Rhodium trifluoroacetate dimer (1 mL of a 1 mg/mL solution in methylenechloride) was added to each of the vessels and the resin was agitated at 25 °C overnight. Each resin batch was filtered then washed separately with methylenechloride (1 x 15 mL). The batches were combined and the entire library (217 compounds) was washed with methylenechloride (5 x 50 mL).

c) Deprotection

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N-Fmoc protecting groups were removed by washing the resin once with DMF, filtering, then suspending the resin in 60 mL 50% piperidine/DMF and agitating at room temperature for 30 min. The resin was filtered, washed (DMF (5 x 50 mL)) and methylenechloride (10 x 50 mL)) and dried in vacuo.

Step 3

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a) Addition of Y

The dried resin from Step 2 (c) was divided into 31 batches of 150 mg (~0.05 mmole), each of which was placed in a 20 mL synthesis vessel. Each of the reagents (0.25 g, > 0.4 mmole, > 8 eq) used in the third step of the synthesis was dissolved in methylenechloride or DMF or a mixture of the two as appropriate (Table 1-3). Each reagent solution was added to one of the 31 resin batches, then any coreagents were added as required (see table). The resin batches were agitated at 25 °C overnight, then each was washed (DMF (1 x 10 mL)) and methylenechloride (10 x 10 mL)).

b) Encoding of Step 3

Stock solutions of 200 mg of C₄Cl₅ and C₃Cl₅ -linker-diazoketone (Preparation 1) were prepared in 16 mL methylenechloride. Stock solutions of 600 mg of C6Cl3, C5Cl3, and C4Cl3 -linker-diazoketone 15 (Preparation 1) were prepared in 16 mL methylenechloride. Aliquots (1 mL) of the stock solutions were used to prepare the appropriate binary coding mixtures for each of the 31 resin batches (Table 1-3). The appropriate coding mixture was added to each of the resin batches and the resin was agitated for 30 min. Rhodium trifluoroacetate dimer 20 (1 mL of a 1 mg/mL solution in methylenechloride) was added to each of the vessels and the resin was agitated at 25 °C overnight. Each resin batch was filtered then washed separately with methylenechloride (1 x 15 mL). The batches were combined and the entire library (6727 compounds) was washed with methylenechloride (5 x 50 mL), then 25 dried in vacuo.

c) Deprotection

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The combined resins from Step 3(b) (~ 2 g) were suspended in 60 mL TFA/thioanisole/EDT (50/50/5) and shaken at room temperature overnight. The resins were filtered, washed (methylenechloride 10 x 50 mL), and dried in vacuo.

d) Decoding Procedure

A bead is placed in a 1.3 mm diameter pyrex capillary with 2 μ L of acetonitrile. Ceric ammonium nitrate solution (2 μ L of a 0.1 M aq.

solution) and hexane (3 μ L) are added and the two-phase mixture centrifuged briefly. The tube is sealed and left at 35 °C for 16 hrs, then opened. The organic layer is removed by syringe and mixed with 1 μ L of N,O-bis(trimethylsilyl)acetamide. The silated tag solution (1 μ L) is analyzed by GC with electron capture (EC) detection.

The GC analysis is performed with a Hewlett Packard 5890 plus gas chromatograph. On column injection into a 5 m, 0.32 mm retention gap connected to a 25 m, 0.2 mm crosslinked 5% phenylmethyl silicone column is used. The temperature and pressure programs for the analysis are 200-320 °C, 15 °C/min, then 320 °C for 10 min and 20-40 psi at 2 psi/min, then 40 psi for 10 min. The EC detector is maintained at 400 °C and the auxiliary gas is set at 35 psi.

Table 1-1
Residues and Encoding Scheme

	R ¹ Residue	Binary Code
5	1. 5-0 N	001
10	2. 5-0 N	010
15	3. 5-0 N	011
	4. 5000NN	100
20	5. 5-0 N ₅	101
25	6. 5-0 Ny	110
30	7. SOCH ₂₎₅ H	111

Table 1-2

<u>R2 Reagents and Encoding Scheme</u>

	Reagent	Binary Code
5	8. (L)	00001
10	9. (D)	00010
15	H-N SMe 10. (D)	00011
20	0'Bu (L)	00100
	H STrityl 12. (L)	00101
25	H STrityl 13. (L)	00110

Table 1-2 (Cont.)

5	H C O₂¹Bu 14. (D)	00111
	H-N-CO ₂ tBu 15. (D)	01000
10	16. (L)	01001
15	17. S—N—S	01010
20	$ \begin{array}{c} $	01011
	H N	01100
25	20. NH-Boc	01101

Table 1-2 (Cont.)

15 23. (D) NH-Mtr NH-Trityl 24. (D) 10000 10001 24. NH-Trityl 25. NH-Trityl 26. NH-Trityl NH-Trityl 10100 10100			
10 $ \begin{array}{cccccccccccccccccccccccccccccccccc$		H	01110
15 23. (D) NH-Mtr NH-Trityl 24. (D) 10000 10001 24. NH-Trityl 25. NH-Trityl 26. NH-Trityl NH-Trityl 10100 10100	. .	H_N NH-Mtr	01111
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10	H NH-Mti	10000
25. NH-Trityl 25. (L) 10011 26 NH-Trityl 27. (D) NH-Trityl 27. (D)	15		10001
25 (L) NH-Trityl 27. (D) 10100	20		10010
NH-Trityl 27. (D)	25	l	10011
90	30	H NH-Trityl	10100

Table 1-2 (Cont.)

15 30. (L) 10111 11000 20 H CH ₃ CH ₃ (D) 11010 25 CH ₃ H CH ₃ CH ₃ CH ₃ CH ₃ 11011			
10 10 10 10 10 10 11 10 11 10 11 10 11 10 11 10 10 11 10 10 11 10 10 11 10 10 11 10 10 11 10 10 10 11 10	5		10101
15 30. (L) 10111 11000 20 H CH ₃ CH ₃ (D) 11010 25 CH ₃ CH ₃ CH ₃ CH ₃ CH ₃ 11011		H-N-CH3	10110
15 31. 20 H CH3 (D) 11000 11001 25 CH3 CH3 CH3 CH3 11011	10	30. (L)	10111
20 $ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	·	T _H ,N	11000
25 CH ₃ 11011		(D)	11001
H-N-CH3	25	H-N-CH ₃	11010
34.		1 N 1	11011

Table 1-2 (Cont.)

_	35. CH ₃ (L)	11100
5	36. CH ₃ (D)	11101
10	37. H-N	11110
15	38.	11111

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Table 1-3
Y Reagents and Encoding Scheme

-54-

5	Y Reagent	Solvent	Co-reagent(s)	Binary Code
10	39. OH S O	DMF	0.3 mL DIC, 100 mg DMAP	00001
15	40. Cl O NH ₂	DMF	0.3 mL DIC, 100 mg DMAP	00010
20	Cl O S-NH ₂ 41. Cl O	DMF	0.3 mL DIC, 100 mg DMAP	00011
25	42. SO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	00100
30	43. SO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	00101

Table 1-3 (Cont.)

		T		·
5	44. CI S, CI	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	00110
10	45. CH ₃ CI	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	00111
	46. O S O	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	01000
15 20	47. SO ₂ Cl NMe ₂	CH ₂ Cl ₂ /DMF (1:1)	0.5 mL NEt3, 100 mg DMAP	01001
	48. n-BuSO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	01010
25	49. (CH ₃) ₃ C —————SO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	01011
	50. H_3C —SO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	01100

Table 1-3 (Cont.)

	51. SO ₂ Cl	CH ₂ Cl ₂	0.5 mL NEt3, 100 mg DMAP	01101
5	52. CO ₂ H	CH ₂ Cl ₂	0.3 mL DIC, 100 mg DMAP	01110
10	53. CO ₂ H	CH ₂ Cl ₂ /DMF (1:1)	0.3 mL DIC, 100 mg DMAP	01111
15	54. CO ₂ H	CH ₂ Cl ₂ /DMF (1:1)	0.3 mL DIC, 100 mg DMAP	10000
20	55. 0 0	CH ₂ Cl ₂	100 mg DMAP	10001
	56. CO ₂ H	CH ₂ Cl ₂ /DMF (1:1)	0.3 mL DIC, 100 mg DMAP	10010
25	57. MeO(CH ₂) ₅ - CO ₂ H	CH ₂ Cl ₂	0.3 mL DIC, 100 mg DMAP	10011
20	58. CO ₂ H	CH ₂ Cl ₂	0.3 mL DIC, 100 mg DMAP	10100
30				

Table 1-3 (Cont.)

	59. H₃C CO ₂ H	CH ₂ Cl ₂	0.3 mL DIC, 100 mg DMAP	10101
5	60. CO ₂ H	CH ₂ Cl ₂ /DMF (1:1)	0.3 mL DIC, 100 mg DMAP	10110
10	O CO ₂ H 61. HN NH	CH2Cl2/DMF (1:1)	0.3 mL DIC, 100 mg DMAP	10111
15	62. N—CO ₂ H	CH ₂ Cl ₂ /DMF (1:1)	0.3 mL DIC, 100 mg DMAP	11000
20	63. F ₃ C NCO	CH ₂ Cl ₂		11001
	64. NCO	CH ₂ Cl ₂		11010
25	65. H ₃ C Cl	CH ₂ Cl ₂	0.5 mL NEt3	11011
	66. CI NCO	CH ₂ Cl ₂		11100
30 .				

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Table 1-3 (Cont.)

1				
5	67. H CO ₂ H	Fmoc-G-OH, DIC, DMAP DMF, then piperidine/ DMF	2-Napthalene- sulfonyl chloride, NEt ₃ , DMAP, CH ₂ Cl ₂	11101
	68. H_CO ₂ H	Fmoc-G-OH, DIC, DMAP DMF, then piperidine/ DMF	4-Biphenyl- carboxylic acid, DIC, DMAP, CH ₂ Cl ₂ /DMF	11110
10	69. N CO ₂ H	Fmoc-G-OH, DIC, DMAP DMF, then piperidine/ DMF	Imidazole- acetic acid, DIC, DMAP, CH ₂ Cl ₂ /DMF	11111
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EXAMPLE 2 DVB-CROSSLINKED VS. PEG-GRAFTED POLYSTYRENE SOLID SUPPORTS

A compound of Formula II of the structure V:

was synthesized essentially as described in Example 1, but without the addition of identifiers. The synthesis was carried out using both 100 μ m 1% DVB-crosslinked polystyrene beads and 130 μ m DVB-crosslinked, PEG-grafted polystyrene beads. At the end of each step, the ligand, or intermediate ligand, was cleaved from a portion of the beads. At the end of Step 3, a portion of the ligand on the PEG-grafted support was

cleaved while still protected and another portion was de-protected prior to cleavage. The ligand on the DVB-crosslinked support was deprotected prior to cleavage. Table 2-1 presents the yield data for both syntheses. The yields were calculated based on the available functionalities (~300 pmole) on each bead.

Table 2-1 Comparative Yields

	DVB Polystyrene (%)	PEG Polystyrene (%)
Step 1	49	85
Step 2	31	45
Step 3 - Protected		58
De-protected	8	61

WHAT IS CLAIMED IS:

1. A compound of the formula:

$$(T'-L)_{q^-}$$
 \bigcirc $-C(O)-L'-II'$

wherein:

5 (S) is a solid support;

T'-L- is an identifier residue;

-L'-II' is a ligand/linker residue; and

q is 3-30.

2. A compound of Claim 1 wherein:

10 T'-L- is of the Formula:

wherein n = 3-12 when Ar is pentachlorophenyl and n = 4-6 when Ar is 2,4,6-trichlorophenyl;

q is 3--L'- is

15

3-13; and

wherein the left-hand bond as shown is the point of attachment to the solid support and the right hand bond is the point of attachment to the ligand, and B is O or NH, with the proviso that in (b) B is NH when attached to a carbonyl group in the ligand.

3. A compound of the formula:

Y-A-CO-R¹ II

wherein:

 R^1 is $-N(R^6)-(CH_2)_{2-5}-Z-(CH_2)_{2-5}-R^9$, $-NH(CH_2)_{2-5}-R^9$, or

$$-N$$
, or $-N$, $(CH_2)_m$, $(CH_2)_m$, R^9 .

R² is

the residue on the α carbon of methionine, O-t-butylserine, serine, S-trityl-cysteine, cysteine, aspartic acid- β -t-butyl ester, aspartic acid, glutamic acid- γ -t-butyl ester, glutamic acid, Nim-trityl-histidine, histidine, NE-Boclysine, lysine, Ng-Mtr-arginine, arginine, N- β -trityl-asparagine, asparagine, N- γ -trityl-glutamine, glutamine, Nin-Boc-tryptophan, tryptophan, isoleucine, phenylalanine, glycine, alanine, valine, or leucine;

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 R^3 is lower alkyl or $-(CH_2)_m$ -Q-X, with the proviso that when m=0, R^{10} is not OH;

D4 :

 R^4 is $-Q(R^7, R^8)-SO_2NH_2$, $-(CH_2)_m-R^{10}$, with the proviso that when m=0, R^{10} is not OH, lower alkyl, a 6-membered aromatic heterocyclic ring containing 1 or 2 N atoms, heteroaryl-lower alkyl,

$$(CH_2)_m$$
 X_{0-2} , $-(CH_2)_m$ X_{0-1} , X_{0-1} ,

R⁵ is lower alkyl, lower cycloalkyl, alkenyl, alkynyl, a mono-

or bicyclic 6- to 10-membered aromatic ring system, or a mono- or bicyclic 5- to 10-membered heteroaromatic ring system containing 1 or 2 N atoms, either system unsubstituted or substituted with 1-2 substituents selected from halogen, alkoxy, alkyl, CF₃, CN, -N(lower alkyl)₂, and acylamino;

10 R⁶ is H or lower alkyl;
R⁷, R⁸ is each independently H, halogen, lower alkyl, alkoxy, CN,
-NO₂, -CO-lower alkyl, -N(lower alkyl)₂, or
NH-CO-lower alkyl;

R⁹ is OH, CONH₂, or COOH;

15 R¹⁰ is alkoxy, OH, or COOH; m is 0-6;

A is -NH-CHR²-, -NH(CH₂)₂₋₁₂-, -N , -N , or the descarboxy residue of a primary or secondary amino acid;

20 Q is a 5- or 6-membered aromatic or heteroaromatic ring containing 0-3 heteroatoms selected from O, N, and S, or a bicyclic 9- or 10-membered aromatic or heteroaromatic ring system containing 0-3 heteroatoms selected from O, N, and S;

X is

H, lower alkyl, halogen, alkoxy, CF₃, CN, -NO₂,
-CO-lower alkyl, -N(lower alkyl)₂, NH-CO-lower alkyl,
or COOH;

Y is -SO₂R³, -COR⁴, -CO-CH(R²)-NHSO₂R³, -CO-CH(R²)-NHCOR⁴, -CO-NHR⁵, or -COOR⁵; and

Z is -O-, -S-, or -N(lower alkyl)-;

or a pharmaceutically acceptable salt thereof.

4. A compound of Claim 3 wherein: Y is

$$H_2NSO_2$$
 C_1 H_2NSO_2 H_2NSO_2 H_2NSO_2 C_1 H_2NSO_2 C_1 H_2NSO_2 C_1 H_2NSO_2 H_2NSO_2 C_1 H_2NSO_2 H

NILL CLUD? NILLOW \ -N \

A is -NH-CHR²-, -NH(CH₂)₂₋₁₂-, -N , -N , or the descarboxy residue of a primary or secondary amino acid; and

wherein R^{11} is CH_3 , $(CH_3)_3C_7$, or Cl;

$$R^1$$
 is

$$-N$$
 $-N$ $-CO_2H$ $-N$ OH $-N$ $-N$ $-OH$ $CONH_2$

5
$$-NH \longrightarrow O \longrightarrow OH$$
 , or $-NH \longrightarrow CO_2H$

5. A compound of Claim 3 wherein: Y is

10
$$SO_2$$
 or $N(CH_3)_2$

15 6. A compound of Claim 3 of the formula:

Y-A-CO-R¹ II

wherein the substituents are as follows:

		to as follows.	
20	Y	Α	R ¹
	1. H ₂ NSO ₂ CO-	-NH CH- (L)	-N—СООН
25	2. H ₂ NSO ₂ CO-	-NHCH- (L)	-N—CH₂OH
30	CI CO- SO ₂ NH ₂	(CH ₂) ₂ -COOH -NH-CH- (L)	-N—CH₂OH

			
	CI CO- 4. CI SO ₂ NH ₂	-NH-(CH ₂) ₅ -	-N_OH
5	5. H ₂ NSO ₂ CO-	-NH CH- (L)	-NH-(CH ₂) ₅ -COOH
10	6. H ₂ NSO ₂ CO-	-NHCH- (L)	$-N$ $-CO_2H$
15	7. CI SO ₂ NH ₂	-NH CH- (L)	$-N$ $-CO_2H$
20	8. CI SO ₂ NH ₂	CH ₃ I -NH-CH- (D)	-NH-(CH ₂) ₅ -COOH
25	9. H ₂ NSO ₂ CO-	-NHCH- (D)	−N—CO ₂ H
30	CI CO- 10. CI SO ₂ NH ₂	CH ₃ -NH·CH- (D)	-NOH

	11. H ₂ NSO ₂ CO-	-NHCH- (D)	-NOH
5	12. H ₂ NSO ₂ CO-	-N_(L)	-N—соон
10	13. H ₂ NSO ₂ CO-	CH(CH ₃) ₂ -NH CH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
15	14. H ₂ NSO ₂ CO-	(CH ₂) ₂ -COOH -NH-CH- (L)	-NH-(CH ₂) ₅ -COOH
20	CI CO- 15. SO ₂ NH ₂	CH ₂ C(O)NH ₂ -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
	16. H ₂ NSO ₂	CH ₂ CO ₂ H -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
25	17. H ₂ NSO ₂ CO-	CH ₂ CO ₂ H -NHCH- (D)	-NCO₂H
30	18. H ₂ NSO ₂	CH(CH ₃) ₂ -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H

	19. H ₂ NSO ₂	CH ₂ CH ₂ SCH ₃ -NHCH- (D)	-N_OH
5	20. H ₂ NSO ₂	-NH(CH ₂) ₆ -	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
10	21. H ₂ NSO ₂	H₃C-CHCH₂CH₃ -NHCH- (L)	-м—-он
15	22. H ₂ NSO ₂	(CH ₂) ₂ CO ₂ H -NHCH- (D)	−N
20	23. H ₂ NSO ₂	CH(CH ₃)₂ -NHCH- (D)	-N—ОН
	24. H ₂ NSO ₂ CO-	CH ₂ CH(CH ₃) ₂ -NHCH- (L)	-мон
25	25. H ₂ NSO ₂ CO-	− Z	-N—ОН
30	26. H ₂ NSO ₂	CH ₂ CH(CH ₃) ₂ -NHCH- (L)	-NH-(CH ₂) ₅ -COOH

	27. H ₂ NSO ₂ CO-	-NHCH- NH (D)	−N CH ₂ OH
5	28. H ₂ NSO ₂ CO-	-NHCH- (L)	−N CH ₂ OH
10	29. H ₂ NSO ₂ CO-	COOH -NHCH- (L)	-м—соон
15	30. H ₂ NSO ₂ CO-	C(O)NH ₂ -NHCH- (L)	-NH-(CH ₂) ₅ -COOH
20	31. H ₂ NSO ₂ CO-	-NH(CH ₂) ₅ -	−NCO ₂ H
	32. H ₂ NSO ₂	CH₃ -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
25	33. H ₂ NSO ₂ CO-	(CH ₂) ₂ CO ₂ H -NHCH- (D)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
30	34. H ₂ NSO ₂ CO-	(CH ₂) ₂ C(O)NH ₂ -NHCH- (D)	-N—CH₂OH

35. H ₂ NSO ₂	(CH ₂) ₂ C(O)NH ₂ -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H
36. H ₂ NSO ₂	(CH ₂) ₂ CO ₂ H -NHCH- (L)	-N-(CH ₂) ₂ -O-(CH ₂) ₂ -OH H

- 7. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 3 and a pharmaceutically acceptable carrier.
- 8. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 4 and a pharmaceutically acceptable carrier
 - 9. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 5 and a pharmaceutically acceptable carrier.
- 10. A pharmaceutical composition comprising a therapeutically effective amount of a compound of Claim 6 and a pharmaceutically acceptable carrier.
- 11. A method of inhibiting serine proteases in a mammal which comprises administering to said mammal an effective amount of a compound of Claim 5.
 - 12. A method of inhibiting carbonic anhydrase isozymes in a mammal which comprises administering to said mammal an effective amount of a compound of Claim 6.
- 13. A method of identifying a ligand having a desired characteristic which comprises synthesizing a combinatorial library of Claim 1 and testing the compounds in said library in an assay which identifies compounds having the desired characteristic.

- 14. A method of Claim 13 further comprising determining the structure of any ligand so identified.
- 15. A method of identifying a ligand having a desired characteristic which comprises synthesizing a combinatorial library of Claim 1, detaching the ligands from the solid supports in said library, and testing said ligands in an assay which identifies compounds having the desired characteristic.
 - 16. A method of Claim 15 further comprising determining the structure of any ligand so identified.
- 17. A process for preparing t-butyl 4-(hydroxymethyl)-3-nitrobenzoate from t-butyl 4-(acetoxymethyl)-3-nitrobenzoate which comprises reacting t-butyl 4-(acetoxymethyl)-3-nitrobenzoate with hydrazine or hydrazine hydrates in a polar solvent and an aprotic polar solvent at 0-50 °C.
- 15 18. The use of divinylbenzene-cross-linked, polyethyleneglycol-grafted polystyrene beads as the solid supports for constructing non-oligomeric combinatorial libraries.
 - 19. A method of Claim 18 wherein the ligand is detached from said solid supports by photolysis.

INTERNATIONAL SEARCH REPORT

lucrnational application No. PCT/US95/03223

	SSIFICATION OF SUBJECT MATTER	
IPC(6) US CL	:Please See Extra Sheet.	
	:Please See Extra Sheet. to International Patent Classification (IPC) or to both national classification and IPC	
	LDS SEARCHED	
	ocumentation searched (classification system followed by classification symbols)	
	Please See Extra Sheet.	
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Documental	ion searched other than minimum documentation to the extent that such documents are included	hin the fields seconded
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Electronic d	ata base consulted during the international search (name of data base and, where practicable	, search terms used)
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C. DOC	IMENTS CONSIDERED TO BE DOVERNOON.	
C. DOC	UMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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Υ	US, A, 5,281,585 (DUGGAN ET AL) 25 JANUARY 1994,	1-12
	COLUMN 3 LINE 1 TO COLUMN 9 LINE 17	·
Y,P	WO A 94/09051 (CTUL ET AL) 44 APRIL 1004 THE	
',''	WO, A, 94/08051 (STILL ET AL) 14 APRIL 1994, PAGE 81.	1-12
y	WO, A, 93/20242 (LERNER ET AL) 14 OCTOBER 1993,	
	PAGE 4 LINES 15+.	1-12
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Y	WO, A, 93/06121 (DOWER ET AL) 01 APRIL 1993, PAGE	1 10
	16 LINES 4+.	1-12
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Furthe	r documents are listed in the continuation of Box C. See patent family annex.	
Spec	ial categories of cited documents: "T" later document published after the inter	mational filing days and in the
A* docu	ment defining the general state of the art which is not considered date and not in conflict with the application of particular relevance principle or theory underlying the inve	ion but cited to understand the
	er document multiplied on or after the international Silve Jan. "X" document of particular relevances the	alaimed investigation and the
L* docu	ment which may throw doubts on priority claim(s) or which is	ed to involve an inventive step
Cited	to establish the publication date of another citation or other al reason (as specified) "Y" document of particular relevance: the	Claimed invention cannot be
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acsimile No.	(703) 305-3230 Telephone No. (703) 308-1235 √210 (second sheet)(July 1992)★	-VO

INTERNATIONAL SEARCH REPORT

l...crnational application No. PCT/US95/03223

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)
This international report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
1. Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely:
Claims Nos.: because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).
Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)
This International Searching Authority found multiple inventions in this international application, as follows: X Please See Extra Sheet.
1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark on Protest The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

n...mational application No. PCT/US95/03223

A. CLASSIFICATION OF SUBJECT MATTER: IPC (6):

A61K 31/095, 38/09, 39/44; C07C 69/06; C07C 211/00, 211/06, 233/19, 233/22, 317/14, 323/22, 323/33,; C07D 207/00, 233/54, 285/10, 307/64, 315/00, 333/34, 333/36; C08F 6/00, 8/00, 8/40, 112/08; C09K 3/00; C12N 1/00; C12P 19/34; C12Q 1/00; G01N 33/53

A. CLASSIFICATION OF SUBJECT MATTER: US CL :

252/182.11; 435/7.92, 91, 960, 973; 514/913; 525/59, 61; 526/341, 347.2; 528/486, 487, 491; 546/216, 219, 221, 226, 227, 229, 233, 235, 236;, 237, 238; 548/135, 311.1, 401, 530, 539,540; 549/3, 64, 201, 479; 560/106; 562/430; 564/154, 162, 305

B. FIELDS SEARCHED

Minimum documentation searched Classification System: U.S.

252/182.11; 435/7.92, 91, 960, 973; 514/913; 525/59, 61; 526/341, 347.2; 528/486, 487, 491; 546/216, 219, 221, 226, 227, 229, 233, 235, 236, 237, 238; 548/135, 311.1, 401, 530, 539, 540; 549/3, 64, 201, 479; 560/106; 562/430; 564/154, 162, 305

BOX II. OBSERVATIONS WHERE UNITY OF INVENTION WAS LACKING This ISA found multiple inventions as follows:

I.Claims 1-2,13-16 drawn to the compound of formula I and a method of using.

II. Claims 3-10 drawn to a composition of the compound of formula II.

III. Claims 11,12 drawn to a method of using the compond of Group II.

IV.Claim 17, drawn to a process for making nitrobenzoate.

V.Claims 18-19 drawn to a method of using polystyrene beads as solid supports.

The inventions listed as Groups I- V do not relate to a single inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons:

There is no technical relationship between the above listed invention groups involving a corresponding special technical feature. Multiple products, processes and uses are claimed and groups of inventions are not so linked as to form a single general inventive concept under 37CFR 1.475.

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